



OFFICE OF NAVAL RESEARCH

Contract N00014-80-C-0472

Task No. NR 056-749

TECHNICAL REPORT No. 24

Laser-Stimulated Adspecies Interaction with a Semiconductor Surface

bу

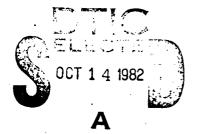
William C. Murphy and Thomas F. George

Prepared for Publication

in

Journal of Physical Chemistry

University of Rochester Department of Chemistry Rochester, New York 14627



September, 1982

Reproduction in whole or in part is permitted for any purpose of the United States Government.

This document has been approved for public release and sale; its distribution is unlimited.

Unclassified +

REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM		
1. REPORT NUMBER	1 .	3. RECIPIENT'S CATALOG NUMBER		
UROCHESTER/DC/82/TR-24	AD-1120259	<u> </u>		
4. TITLE (and Subtitio)		5. TYPE OF REPORT & PERIOD COVERED		
Laser-Stimulated Adspecies Interaction with a Semiconductor Surface		Interim Technical Report		
		5. PERFORMING ORG. REPORT NUMBER		
7. AUTHOR(s) William C. Murphy and		B. CONTRACT OR GRANT NUMBER(#)		
Thomas F. George		N00014-80-C-0472		
9. PERFORMING ORGANIZATION NAME AND ADDRESS		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS		
University of Rochester Department of Chemistry		NR 056-749		
Rochester, New York 14627		NK 030-743		
11. CONTROLLING OFFICE NAME AND ADDRESS		12. REPORT DATE		
Office of Naval Research		September 1982		
Chemistry Program Code 472 Arlington, Virginia 22217		13. NUMBER OF PAGES		
14. MONITORING AGENCY NAME & ADDRESS(II dillorent	from Controlling Office)	15. SECURITY CLASS. (of this report)		
		Unclassified		
		15a DECLASSIFICATION/DOWNGPADING		
		15. DECLASSIFICATION/DOWNGRADING SCHEDULE		
This document has been approved for public release and sale; its distribution is unlimited.				
17. DISTRIBUTION STATEMENT (of the obstract entered in Block 20, if different from Report)				
				
Prepared for publication in Journal of Physical Chemistry, in press.				
19. KEY WORDS (Continue on reverse elds if necessary and identify by block number) SURFACE STATES CHARGE DISTRIBUTION SEMICONDUCTOR ENHANCED ADSPECIES-SURFACE INTERACTION SILICON ONE-DIMENSIONAL MODEL LASER EXCITATION QUANTUM MECHANICAL				
The surface electronic charge distributions of a one-dimensional semi-conductor are compared for the ground state and the laser-excited surface states. A charged adspecies interaction potential with these excited surface states is examined for the case of silicon. The use of a laser to enhance desorption or adsorption by this process is discussed.				

Laser-Stimulated Adspecies

Interaction with a

Semiconductor Surface

William C. Murphy

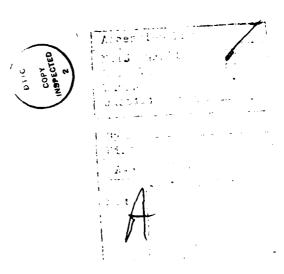
and

Thomas F. George

Department of Chemistry
University of Rochester
Rochester, New York 14627

Abstract

The surface electronic charge distributions of a one-dimensional semiconductor are compared for the ground state and the laser-excited surface states. A charged adspecies interaction potential with these excited surface states is examined for the case of silicon. The use of a laser to enhance desorption or adsorption by this process is discussed.



1. Introduction

Laser-induced photochemistry with adspecies is a growing area of interest, with much attention being given to desorption processes. For example, Djidjoev and co-workers¹ have reported that the desorption of hydroxyl groups from the surface of silica can be greatly enhanced in the presence of an infrared laser. Chuang² has also presented experimental evidence that the reaction of an adspecies with a silicon surface can be stimulated by laser radiation.

For the theoretical descriptions of these processes, a statistical analysis has been given by Lin et al., while Jedrzejck and his associates have used a simple one-dimensional model. Both of these theoretical studies, and the experimental works, have relied on the use of the laser to excite the vibrational modes of the system.

On the other hand, photo-induced surface reactions can occur through electronic excitation. Such effects have been demonstrated by synchrotron radiation studies⁵ on metal surfaces. Desorption of an adspecies is produced by a shift in electronic charge in the surface region and the resultant Coulombic repulsion.⁶

For a semiconductor with a surface, the electronic energies consist of various surface states, in addition to the bulk valence and conduction bands. These surface states, with charge localized in the surface region, can be populated by using a laser to excite electrons from the bulk electronic bands. As in the case of synchrotron radiation, such surface charge could induce a strong

Coulombic interaction with an adspecies.

In the next section, we calculate the electronic density profile of a model one-dimensional semiconductor, after which the effect of surface state excitation on this curve is examined. The changes in the interaction potential of a charged adspecies with the surface in various excited states is then calculated and compared. Finally, limitations and possible extensions of our model and its application to laser-induced surface processes are discussed.

2. Electronic Density Profile

In our model, we have a one-dimensional linear chain of atoms of length L and with lattice constant a, which is confined along the z-axis to the region z < a/2. Such a model has been shown to produce valence, conduction and surface bands. 7,8 In the ground state, the valence band is completely occupied while the surface and conduction bands are empty.

To obtain the total electron density, n(z), we must sum the individual densities of all occupied states:

$$n(z) = \frac{8L}{(2\pi)^3} \int d^3k |\psi_k(z)|^2$$
 (1)

where the wave function, $\psi_k(z)$, with crystal momentum k was previously determined. 7,8 If we now use cylindrical coordinates, this can be readily reduced to

$$n(z) = \frac{2L}{\pi^2} \int_0^{k_F} dk [E_F - E(k)] |\psi_k(z)|^2$$
 (2)

where $\mathbf{E}_{\mathbf{F}}$ is the Fermi energy of crystal momentum $\mathbf{k}_{\mathbf{F}}$.

Because of its importance as a semiconductor, we have used parameters^{9,10} typical of silicon to numerically evaluate the density profile, and our results are shown by the solid line in Figure 1. The oscillations of the charge as one goes into the bulk of the crystal and the exponential tail are typical of such density profiles. These results are for the ground electronic state. The effect on this profile of promoting electronic charge by laser action into the surface states is considered below.

3. Charge Density of the Excited System

If the system is now exposed to a laser with an energy of about half the band gap, it is possible to populate surface states of our crystal. The wave function of these surface states is of the form

$$\psi_{\kappa} = C_{\rm s} e^{-\kappa (z - \frac{a}{2})} \sin \left[\frac{g}{2} (z - \frac{a}{2}) + \theta_{\kappa} \right]$$
 (3a)

for $z < \frac{a}{2}$ and

$$\psi_{\kappa} = C_{\mathbf{S}} \sin \theta_{\kappa} e^{-\mathbf{q}(\mathbf{z} - \frac{\mathbf{a}}{2})}$$
(3b)

for $z > \frac{a}{2}$, with the energy given by

$$E(\kappa) = \frac{1}{2} [(\frac{g}{2})^2 - \kappa^2 \pm (4V^2 - \kappa^2 g^2)^{1/2}]$$
 (4)

where the normalization constant, C_g , the exponential factor, q, and the phase factor, θ_K , have previously been determined. The parameter κ in the above expressions is the imaginary part of the crystal momentum that occurs in surface states. V, the g component of the Fourier transform of the effective potential where $g=2\pi/a$ is the reciprocal lattice constant, is assumed to be positive.

Using Equation (1) we find the new charge density to be

$$n(z) = n_0(z) - |\psi_{\kappa}(z)|^2 + |\psi_{\kappa}(z)|^2,$$
 (5)

where $n_0(z)$ refers to the charge density of the unexcited system given by Equation (2) and the transition proceeds from bulk state k to surface state κ . As previously pointed out, 8 the excitation conserves the crystal momentum. Therefore, bulk states are excited at the band edge with $k=\frac{g}{2}$:

$$\psi_{\underline{q}}(z) = \frac{2}{L^{1/2}} \sin \left[\frac{q}{2} (z - \frac{a}{2}) + \theta_{\underline{q}} \right]. \tag{6}$$

However, since the charge associated with this state goes as 1/L, for a very large system we would in effect be taking only an infinitesimal amount of charge from everywhere in the semiconductor to populate the surface state. Consequently, for a large system, the new charge density can be written

$$n(z) = n_0(z) + |\psi_{\kappa}(z)|^2.$$
 (7)

Using Equation (3) in Equation (7) along with the results for the ground state, we have obtained the density of the semiconductor with the $\kappa = -.5(2V/g)$ lower branch surface state excited. The results are depicted in Figure 1 by the dashed line. As can be seen by the plot, this excited state produces an electronic charge in the surface region that is twice as great as the bulk average. If the surface states are excited closer to the branch point near the gap center, $\kappa = -2V/g$, the charge concentration in the first few layers of the surface will increase up to about thrice the average density. The interaction of an adspecies with this charged surface will now be examined.

4. Adspecies-Surface Interaction

If there is a charged adspecies above our surface, the classical interaction between this ion and our semiconductor can be written as

$$U(z_{I}) = -\int n(z) v(r) dr^{\dagger}$$
(8)

with

$$r = [x^2 + y^2 + (z - z_I)^2]^{1/2},$$
 (9)

where v(r) is the electron-ion potential of the adspecies at $z_{\rm I}$. Let us assume that v(r) is Coulombic in nature with Thomas-Fermi screening: 11

$$v(r) = \frac{Ze^{-\lambda r}}{r}$$
 (10)

where λ is the screening parameter,

$$\lambda^2 = \frac{6\pi \overline{n}}{E_F}, \qquad (11)$$

and Z is the charge or the adspecies. A more appropriate screening may well be similar to those developed for finite metals, 12 but such a calculation is beyond the scope of this paper. Since the Thomas-Fermi screening parameter depends on the average electron density, \overline{n} , we would expect this parameter to be less in the surface region. Consequently, the use of a screening parameter based on the bulk density would underestimate the actual interaction.

Bearing this limitation in mind, Equations (8) and (10) can be combined and readily simplified to give the interaction

$$U(z_{I}) = -\frac{2\pi Z}{\lambda} \int_{-\infty}^{\infty} n(z) e^{-\lambda |z-z_{I}|} dz.$$
 (12)

Inserting Equation (7) into this expression, we obtain

$$U(z_{I}) = -\frac{2\pi z}{\lambda} \int_{-\infty}^{\infty} n_{0}(z) e^{-\lambda |z-z_{I}|} -S_{U(z_{I})} dz, \qquad (13)$$

where the change in potential induced by the excited surface states is given by

$$S_{\mathbf{U}}(z_{\mathbf{I}}) = \frac{2\pi z}{\lambda} \int_{-\infty}^{\infty} |\psi_{\kappa}|^2 e^{-\lambda |\mathbf{z} - \mathbf{z}_{\mathbf{I}}|} dz.$$
 (14)

Inserting the expression for the surface wave function, we get

$$S_{U}(z_{I}) = \frac{2\pi z c_{S}^{2}}{\lambda} \left\{ \int_{-\infty}^{a/2} \sin^{2}\left[\frac{q}{2}(z-\frac{a}{2}) + \theta_{K}\right] e^{-2\kappa(z-\frac{a}{2}) - \lambda |z-z_{I}|} dz + \sin^{2}\theta_{K} \int_{a/2}^{\infty} e^{-2q(z-\frac{a}{2}) - \lambda |z-z_{I}|} dz \right\}.$$
(15)

After much algebra, these integrals yield an interaction potential of the form

$$\frac{S_{U(z_{\underline{I}})}}{Z} = e^{-\lambda z_{\underline{I}}} A(\kappa) - e^{-2qz_{\underline{I}}} B(\kappa), \qquad (16)$$

where

$$A(\kappa) = \frac{C_S^2 \pi}{\lambda} \left[\frac{\sin^2 \theta_{\kappa}}{(q - \frac{\lambda}{2})} - \frac{1}{2\kappa - \lambda} + \frac{(2\kappa - \lambda)\cos 2\theta_{\kappa} - g\sin 2\theta_{\kappa}}{(2\kappa - \lambda)^2 + g^2} \right] e^{\lambda a/2}$$
 (17)

and

$$B(\kappa) = \frac{C_S^2 4 \pi \sin^2 \theta}{(2\sigma)^2 - \lambda^2} \epsilon^{qa}.$$
 (18)

Equation (16) has been evaluated for a number of surface states, and the results are plotted in Figure 2. The curves clearly show that as one moves to larger $|\kappa|$ (energies near the gap center), both the intensity and the range of the surface charge interaction increase, as would be expected from the density calculation. All curves, however, show an appreciable contribution to the potential produced by the surface states with $|\kappa| > 0.1(2V/g)$.

A better comparison of the surface charge interaction among the various surface states is given in Figure 3, where we have plotted the change in potential at z_T = a for all the surface states.

The upper branch states are at a higher energy [positive sign in Equation (4)] than the lower branch states. Therefore, the exponential tail and, subsequently, the interaction is slightly greater.

5. Discussion

Use of a laser to localize electronic charge in the surface region of a semiconductor can produce an appreciable effect on an adspecies-surface interaction. If the adspecies is negatively charged, desorption can be induced; if positively charged, the possibility of adsorption is enhanced. In a more realistic model with both occupied and empty surface states, the laser excitation of holes as well as electrons could be used to selectively enhance adsorption or desorption for the same charged species.

Since the concentration of charge is so great in the surface region, as shown in Figure 1, one would expect the effective interaction length to be greater than that indicated by Figure 2. The exponential decay of our potential is probably an artifact of the assumed Thomas-Fermi screening. An improved interaction potential would necessitate addressing the dielectric screening problem in more detail. 12

Another consideration is the fact that we did not look at the equilibrium concentration of the surface state - only one surface state was assumed to be populated. On the other hand, several partially occupied surface states may well have as much as or more surface charge than a single occupied surface state.

Of course, the major limitation of our model is its onedimensionality. The three-dimensional interaction potential may be quite complex depending not only on the distance from the surface but also on the position of the adspecies with respect to the plane of the surface. Also, many of the features of a crystal cannot be adequately treated in a one-dimensional model.

Nonetheless, our contention that lasers used to control surface charge density can lead to enhanced surface processes has been substantiated. The effects on this process of higher dimensions, phonons, and the dynamics of the adspecies and surface are the subject of continuing research.

Acknowledgments

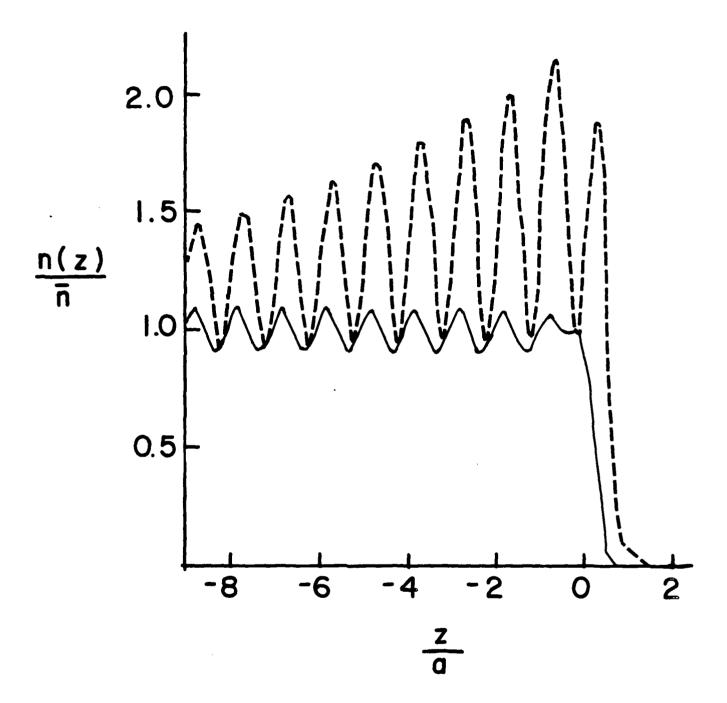
This work was supported in part by the Office of Naval Research and the Air Force Office of Scientific Research (AFSC), United States Air Force, under Grant AFOSR-82-0046. TFG acknowledges the Camille and Henry Dreyfus Foundation for a Teacher-Scholar Award (1975-1982).

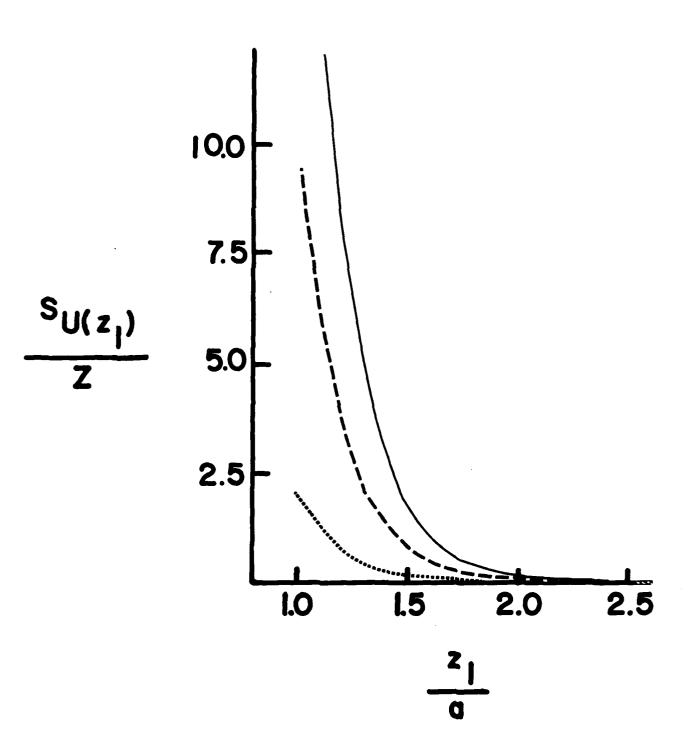
- M. S. Djidjoev, R. V. Knokhlov, A. V. Kiselev, V. I. Lygin,
 V. A. Namiot, A. I. Osipov, V. I. Panchenko and B. I. Provotorov,
 in Tunable Lasers and Applications, edited by A. Mooradian,
 T. Jaeger and P. Stokseth (Springer, Berlin, 1976).
- 2. T. J. Chuang, J. Chem. Phys. 74, 1453 (1981).
- 3. J. Lin, A. C. Beri, M. Hutchinson, W. C. Murphy and T. F. George, Phys. Lett. 79A, 233 (1980).
- 4. C. Jedrzejek, K. F. Freed, S. Efrima and H. Metiu, Surface Sci. 109, 191 (1981).
- D. P. Woodruff, M. M. Traum, H. H. Farrell, N. V. Smith, P. D. Johnson, D. A. King, R. L. Benbow and Z. Hurych, Phys. Rev. B 21, 5642 (1980); R. Jaeger, J. Feldhaus, J. Haase, J. Stöhr, Z. Hussain, D. Menzel and D. Norman, Phys. Rev. Lett. 45, 1870 (1980).
- 6. M. L. Knotek and P. J. Feibelman, Phys. Rev. Lett. 40, 964 (1978).
- 7. S. Lundqvist, in <u>Surface Science</u>, Vol. 1 (International Atomic Energy Agency, Vienna, 1975), p. 331.
- 8. W. C. Murphy and T. F. George, Surface Sci. 114, 189 (1982).
- 9. C. Kittel, Introduction to Solid State Physics, 4th ed. (Wiley, New York, 1971), p. 38, 364.
- 10. J. Van Laar and J. Scheer, Philips Res. Rept. 17, 101 (1962).
- 11. See, e.g., C. Kittel, Quantum Theory of Solids (Wiley, New York, 1963), p. 105.
- 12. D. M. Newns, Phys. Rev. B 1, 3304 (1970); E. Gerlach, in Molecular Processes on Solid Surfaces, E. Dranglis, R. Gretz, and R. Jaffee, eds. (McGraw-Hill, 1969), p. 181.

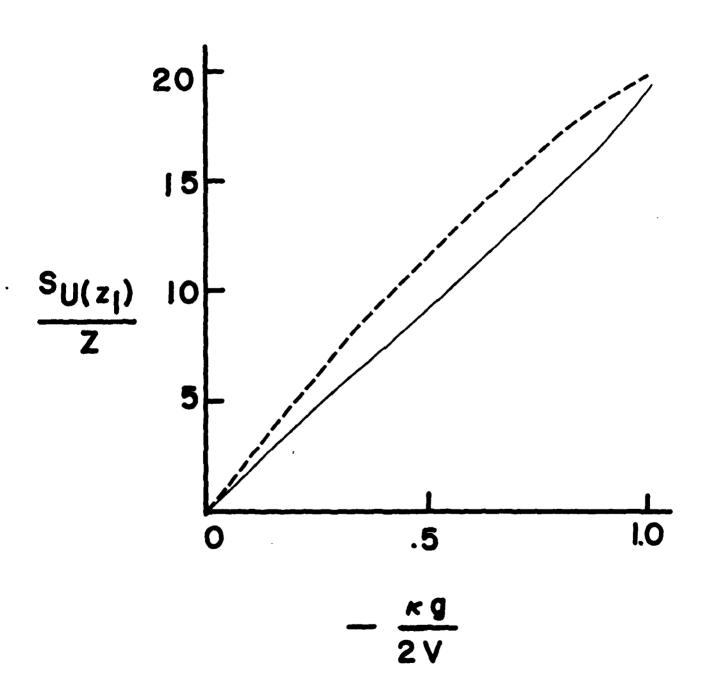
Figure 1. Electron density distribution at the surface. The solid line represents the ground electronic state, and the dashed line represents the system with the excited surface state $\kappa = -.5$ (2V/g) in the lower branch.

Figure 2. The magnitude of the surface interaction potential (in millihartrees) at various distances from the surface. The solid line represents the system with excited state $\kappa \approx -(2V/g)$; the dashed line, $\kappa = -.5(2V/g)$; and the dotted line, $\kappa = -.1(2V/g)$, all in the lower energy branch.

Figure 3. The magnitude of the surface interaction potential (in millihartrees) at a distance $z_{\rm I}$ = a for the system with various excited surface states. The solid line represents surface states in the lower energy branch; the dashed line, the upper energy branch.







TECHNICAL REPORT DISTRIBUTION LIST, GEN

	No. Copies		No.
	Cobies		Copies
Office of Naval Research		Naval Ocean Systems Center	
Attn: Code 413		Attn: Mr. Joe McCartney	
800 North Quincy Street		San Diego, California 92152	1
Arlington, Virginia 22217	2	-	
		Naval Weapons Center	
ONR Pasadena Detachment		Attn: Dr. A. B. Amster,	
Attn: Dr. R. J. Marcus		Chemistry Division	
1030 East Green Street		China Lake, California 93555	, 1
Pasadena, California 91106	l		
		Naval Civil Engineering Laboratory	
Commander, Naval Air Systems Command		Attn: Dr. R. W. Drisko	
Attn: Code 310C (H. Rosenwasser)		Port Hueneme, California 93401	1
Department of the Navy			
Washington, D.C. 20360	1	Dean William Tolles	
		Naval Postgraduate School	
Defense Technical Information Center		Monterey, California 93940	1
Building 5, Cameron Station			
Alexandria, Virginia 22314	12	Scientific Advisor	
		Commandant of the Marine Corps	
Dr. Fred Saalfeld		(Code RD-1)	
Chemistry Division, Code 6100		Washington, D.C. 20380	1
Naval Research Laboratory			
Washington, D.C. 20375	1	Naval Ship Research and Development Center	
U.S. Army Research Office		Attn: Dr. G. Bosmajian, Applied	
Attn: CRD-AA-IP		Chemistry Division	
P. O. Box 12211		Annapolis, Maryland 21401	1
Research Triangle Park, N.C. 27709	1		-
,	_	Mr. John Boyle	
Mr. Vincent Schaper		Materials Branch	
DTNSRDC Code 2803		Naval Ship Engineering Center	
Annapolis, Maryland 21402	1	Philadelphia, Pennsylvania 19112	1
Naval Ocean Systems Center		Mr. A. M. Anzalone	
Attn: Dr. S. Yamamoto		Administrative Librarian	
Marine Sciences Division		PLASTEC/ARRADCOM	
San Diego, California 91232	1	Bldg 3401	
		Dover, New Jersey 07801	1
Dr. David L. Nelson			
Chemistry Program			
Office of Naval Research			
800 North Quincy Street	_		
Arlington, Virginia 22217	1		

TECHNICAL REPORT DISTRIBUTION LIST, 056

	No.		No.
<u>c</u>	opies		Copies
Dr. G. A. Somorjai		Dr. W. Kohn	
Department of Chemistry		Department of Physics	
University of California		University of California	
Berkeley, California 94720	1	(San Diego)	
Beigerey, Odillottata 74725	•	La Jolla, California 92037	1
Dr. J. Murday			_
Naval Research Laboratory		Dr. R. L. Park	
Surface Chemistry Division (6170)		Director, Center of Materials	
455 Overlook Avenue, S.W.		Research	
Washington, D.C. 20375	1	University of Maryland	
		College Park, Maryland 20742	}
Dr. J. B. Hudson			
Materials Division		Dr. W. T. Peria	
Rensselaer Polytechnic Institute		Electrical Engineering Department	
Troy, New York 12181	1	University of Minnesota	
		Minneapolis, Minnesota 55455	J
Dr. Theodore E. Madey			
Surface Chemistry Section		Dr. Chia-wei Woo	
Department of Commerce		Department of Physics	
National Bureau of Standards		Northwestern University	
Washington, D.C. 20234	1	Evanston, Illinois 60201	1
Dr. J. M. White		Dr. Robert M. Hexter	
Department of Chemistry		Department of Chemistry	
University of Texas		University of Minnesota	
Austin, Texas 78712	1	Minneapolis, Minnesota 55455	1
De Polch II Johnson		Dr. B. B. Von Burne	
Dr. Keith H. Johnson		Dr. R. P. Van Duyne Chemistry Department	
Department of Metallurgy and Materials Science		Northwestern University	
		Evanston, Illinois 60201	1
Massachusetts Institute of Technology Cambridge, Massachusetts 02139	1	Evanseon, IIIInois octor	•
Campitage, massachusetts 02139	•	Dr. S. Sibener	
Dr. J. E. Demuth		Department of Chemistry	
IBM Corporation		James Franck Institute	
Thomas J. Watson Research Center		5640 Ellis Avenue	
P. O. Box 218		Chicago, Illinois 60637	1
Yorktown Heights, New York 10598	1		_
	-	Dr. M. G. Lagally	
Dr. C. P. Flynn		Department of Metallurgical	
Department of Physics		and Mining Engineering	
University of Illinois		University of Wisconsin	
Urbana, Illinois 61801	1	Madison, Wisconsin 53706	1
	-	•	

TECHNICAL REPORT DISTRIBUTION LIST, 056

	No. Copies		No. Copies
Dr. Robert Gomer		Dr. K. G. Spears	
Department of Chemistry		Chemistry Department	
James Franck Institute		Northwestern University	
5640 Ellis Avenue		Evanston, Illinois 60201	1
Chicago, Illinois 60637	1		
		Dr. R. W. Plummer	
Dr. R. G. Wallis		University of Pennsylvania	
Department of Physics		Department of Physics	
University of California, Irvine	•	Philadelphia, Pennsylvania 19104	1
Irvine, California 92664	1	n	
Dr. D. Barrakarı		Dr. E. Yeager	
Dr. D. Ramaker		Department of Chemistry	
Chemistry Department		Case Western Reserve University	•
George Washington University	•	Cleveland, Ohio 41106	1
Washington, D.C. 20052	1		
Box B. Harrison		Professor D. Hercules	
Dr. P. Hansma		University of Pittsburgh	
Physics Department		Chemistry Department	•
University of California,		Pittsburgh, Pennsylvania 15260	1
Santa Barbara	•	To B. 18. 18. 18. 18. 18. 18. 18. 18. 18. 18. 18.	
Santa Barbara, California 93106	1	Professor N. Winograd	
Dr. 1 G. Harridanan		The Pennsylvania State University	
Dr. J. C. Hemminger		Department of Chemistry	
Chemistry Department		University Park, Pennsylvania 16802	1
University of California, Irvine Irvine, California 92717	1	Brofoson T. B. Com	
irvine, Calliornia 92/1/	1	Professor T. F. George The University of Rochester	
Dr. Martin Fleischmann		Chemistry Department	
Department of Chemistry		Rochester, New York 14627	1
Southampton University		ROCHESCEL, NEW TOTA 14027	•
Southampton \$09 5NH		Professor Dudley R. Herschbach	
Hampshire, England	1	Harvard College	
nembanzie, andzene	•	Office for Research Contracts	
Dr. G. Rubloff		1350 Massachusetts Avenue	
IRM		Cambridge, Massachusetts 02138	1
Thomas J. Watson Research Center		odimilage, massachasetts varso	•
P. O. Box 218		Professor Horia Metiu	
Yorktown Heights, New York 10598	1	University of California,	
	-	Santa Barbara	
Dr. J. A. Gardner		Chemistry Department	
Department of Physics		Santa Barbara, California 93106	1
Oregon State University			~
Corvallis, Oregon 97331	1	Professor A. Steckl	
	-	Rensselaer Polytechnic Institute	
Dr. G. D. Stein		Department of Electrical and	
Mechanical Engineering Department		Systems Engineering	
Northwestern University		Integrated Circuits Laboratories	
Evanston, Illinois 60201	1	Troy, New York 12181	1
, · · · · · · · · · · · · · · ·	-	,,	•

TECHNICAL REPORT DISTRIBUTION LIST, 056

	No. Copies	No. Copies
Dr. John T. Yates Department of Chemistry University of Pittsburgh Pittsburgh, Pennsylvania 15260	1	•
Professor G. H. Morrison Department of Chemistry Cornell University		•
Ithaca, New York 14853	1	
Captain Lee Myers AFOSR/NC Bolling AFB		
Washington, D.C. 20332	1	
Dr. David Squire Army Research Office P. O. Box 12211		
Research Triangle Park, NC 27709	1 .	
Professor Ronald Hoffman Department of Chemistry Cornell University		,
Ithaca, New York 14853	1 .	